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Key indicators

Single-crystal X-ray study T = 193 K Mean σ (C–C) = 0.004 Å R factor = 0.036 wR factor = 0.086 Data-to-parameter ratio = 7.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{27}H_{40}O_6$, contains two fused cyclohexane rings with chair conformations, fused to a norbornene ring system. The ester groups that are attached to the norbornene unit are in a *cis* configuration. There are two molecules in the asymmetric unit. The structure is stabilized by $C-H \cdots O$ intermolecular hydrogen bonds and van der Waals forces.

Maleopimaric acid trimethyl ester

Comment

Maleopimaric acid (Zalkov *et al.*, 1962) is a derivative of abietic acid, an important resin acid (Lee *et al.*, 2001) used in materials processing, such as in surface active agents and paints. It is also a useful intermediate for natural product synthesis (Khlebnikova *et al.*, 2004).



The title compound, (I), is an ester derivative of maleopimaric acid (Fig. 1). The asymmetric unit consists of two independent molecules. Each molecule contains two fused cyclohexane rings with chair conformations and a fused norbonene ring system. The C1–C6 and C28–C33 sixmembered rings display boat conformations. The H atoms on atoms C1, C2, C5 and C11 in the first molecule and C28, C29, C32 and C38 in the second molecule occupy axial positions, which are similar to those in a derivative of abietic acid (Li *et al.*, 2005). The ester groups at C1/C2 and C28/C29 are in *cis* positions as expected (Zalkov *et al.*, 1962). The configuration about the C7–C8 and C34–C35 bonds in the two molecules is *Z*, where the H atom and the isopropyl group are *cis*. All the bond lengths and angles are in normal ranges (Allen *et al.*, Received 24 June 2006 Accepted 15 November 2006

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1987). In the crystal structure, the molecules are linked (Fig. 2) by $C-H\cdots O$ intermolecular hydrogen bonds (Table 2) and van der Waals forces.

Experimental

Maleopimaric acid was prepared according to a literature procedure (Zalkov *et al.*, 1962). To a 50 ml three-necked flask were added maleopimaric acid (4.0 g, 4 mmol) and phosphorus trichloride (3 ml, 34.3 mol). The mixture was stirred vigorously with a magnetic stirrer at room temperature for 4 h. After removing the solvent under vacuum, anhydrous methanol (20 ml) was added to the residue and refluxed for 3 h. The mixture was cooled to room temperature and then filtered. The colorless solid formed was recrystallized from ethanol (2.9 g, yield 63%, m.p. 375–377 K).

Z = 4

 $D_r = 1.240 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.09 \text{ mm}^{-1}$

T = 193 (2) K

Plate, colorless

 $R_{\rm int}=0.026$

 $\theta_{\rm max} = 25.3^{\circ}$

 $0.79 \times 0.60 \times 0.12~\mathrm{mm}$

24150 measured reflections

 $w = 1/[\sigma^2(F_0^2) + (0.0442P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.5815P]

 $\Delta \rho_{\rm max} = 0.41 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.16 \text{ e} \text{ Å}^{-3}$

 $(\Delta/\sigma)_{\rm max} < 0.001$

4615 independent reflections

4475 reflections with $I > 2\sigma(I)$

Crystal data

 $\begin{array}{l} C_{27}H_{40}O_6\\ M_r=460.59\\ Monoclinic, P2_1\\ a=8.9848~(15)~\text{\AA}\\ b=23.456~(4)~\text{\AA}\\ c=12.364~(2)~\text{\AA}\\ \beta=108.804~(3)^\circ\\ V=2466.6~(7)~\text{\AA}^3 \end{array}$

Data collection

Rigaku Mercury diffractometer ω scans Absorption correction: multi-scan (Jacobson, 1998) $T_{\min} = 0.804, T_{\max} = 0.990$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.036$ $wR(F^2) = 0.086$ S = 1.074615 reflections 610 parameters H-atom parameters constrained

Table 1

Selected bond lengths (Å).

O1-C17	1.205 (3)	O5-C25	1.201 (3)
O2-C17	1.348 (3)	O6-C25	1.342 (3)
O3-C19	1.203 (4)	C7-C8	1.330 (3)
O4-C19	1.344 (3)		

Tal	ble	2
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Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C2-H2\cdots O11^{i}$	1.00	2.47	3.357 (3)	147
C18−H18C···O3 ⁱⁱ	0.98	2.56	3.377 (4)	141
$C26-H26C\cdots O1^{iii}$	0.98	2.51	3.389 (4)	149
C29-H29···O5	1.00	2.49	3.386 (3)	150
$C53-H53C\cdots O9^{iv}$	0.98	2.60	3.513 (4)	156

Symmetry codes: (i) x - 1, y, z; (ii) x + 1, y, z; (iii) $-x + 1, y + \frac{1}{2}, -z + 1$; (iv) $-x + 2, y - \frac{1}{2}, -z + 2$.

H atoms were positioned geometrically and included in the refinement in the riding-model approximation, with C-H = 0.95, 0.98, 0.99 and 1.00 Å for aromatic, methyl, CH_2 and CH groups,



Figure 1

The structures of the two independent molecules in the asymmetric unit of (I), showing the atom-labeling scheme and 50% probability displacement ellipsoids. H atoms are represented by small spheres of arbitrary radii.



Figure 2

Packing diagram for (I). Hydrogen bonds are denoted by dashed lines.

respectively, and with $U_{iso}(H) = 1.5U_{eq}(\text{methyl C})$ and $1.2U_{eq}(\text{other C})$. In the absence of significant anomalous scattering effects, Friedel pairs were merged.

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSC, 2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 2000); software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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